

N-LIMITED VERSUS Ga-LIMITED GROWTH ON GaN(000 $\bar{1}$) BY MBE USING NH₃

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GaN was grown on GaN(000 $\bar{1}$) by MBE using NH₃ and a Ga Knudsen cell. The growth kinetics on samples of this polarity were investigated with desorption mass spectroscopy (DMS) and reflection high energy electron diffraction (RHEED). Both techniques were used to observe and control surface termination, Ga condensation and surface temperature. GaN growth and decomposition rates were measured by DMS. Two stable surface terminations were found to exist — N-terminated and Ga-terminated GaN(000 $\bar{1}$). The N-terminated surface also contained hydrogen which desorbed during growth at a rate proportional to the growth rate. Low temperature reconstructions were only observed by adding weakly adsorbed Ga on top of the Ga-terminated surface. During growth two distinct growth regimes were identified: growth under excess NH₃ and growth under excess Ga. Growth is limited in both regimes by GaN decomposition at high temperatures with an activation energy of 3.4 eV. Growth in the excess Ga regime ceased below the Ga condensation temperature. Under conditions of excess NH₃, strong but damped oscillations in the specular RHEED intensity were observed on smooth surfaces. Contrary to previous suggestions, the period of these oscillations did not correspond exactly to integral layer deposition and was not characteristic of a narrow growth front. Further, the growth mode changed from island nucleation to step flow with an activation energy of 1.2 eV. Under conditions of excess Ga, the diffraction was 2-D but RHEED intensity oscillations were not observed, indicating a step flow growth mode. In this latter regime RHEED measurements were very sensitive to termination changes on the GaN(000 $\bar{1}$) surface, and the growth rate was found to decrease linearly with increasing Ga flux. This reduction is explained by a model in which weakly adsorbed Ga blocks reaction at strongly bound Ga. A map is presented to provide a framework for categorizing the overall growth.

1. Introduction

GaN is a direct, wide band gap semiconductor, and is therefore attractive for blue LED's and lasers, as well as high temperature and power devices.¹ Since the mid-80's, major efforts have been made to produce high quality material using a variety of growth methods.^{2,3} The growth has proven difficult. Among the multiple challenges in growing high quality material by molecular beam epitaxy (MBE) is the control of growth kinetics, which require a good under-

standing of the processes taking place on the GaN surface. In order to study growth kinetics the accurate determination of growth conditions is essential, including crystal polarity, fluxes and surface temperature. Knowing these parameters permits us to investigate their effect on surface termination, growth rate and growth mode.

For the growth of GaN by MBE using NH₃, the kinetics are complicated by (1) the temperature dependence of the dissociation and incorporation of

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NH_3 and (2) the temperature and compositional dependence of the residence times of the reactants on the GaN surface. This means that without knowledge of the microscopic surface processes it is not clear how to set optimum growth parameters. Since the flux and temperature dependencies of the Ga, N and H surface coverages are not well known, the species that limit the growth cannot be easily identified. To monitor these growth processes we have measured structural and compositional changes at the surface with RHEED, and changes in the species leaving the surface with desorption mass spectroscopy (DMS). These two techniques complement each other in determining microscopic processes.

A number of issues related to the growth kinetics of GaN must be addressed. Historically, the n -type conductivity commonly exhibited by GaN was attributed to N vacancies.⁴ This problem was thought to arise from a kinetic barrier to N incorporation during growth. Results by Lee *et al.*⁵ as well as Jones *et al.*⁶ indicate that the GaN growth rate is strongly temperature-dependent, exhibiting a maximum between 750°C and 800°C. The high temperature decrease in the growth rate was attributed to Ga desorption at elevated substrate temperature^{5,6} as well as to the decomposition of GaN.⁷ The common belief at low substrate temperatures is that reduced N incorporation efficiency limits the growth rate.⁵

In this paper we will review the growth kinetics on hexagonal films identified as GaN(000 $\bar{1}$),⁸⁻¹⁰ and show that there are at least two distinct regimes of growth.¹¹⁻¹³ These regimes show different surface morphologies and growth dynamics, primarily due to the surface coverages of Ga and nitrogen, along with hydrogen.^{14,15} The main result is that the diffusion length of adsorbed Ga is much smaller under conditions in which the Ga flux limits the rate of growth. However, it is not yet apparent which growth mode is optimal for doping or for growth on strain-free surfaces.

In the next section the instrumentation is described, the sample preparation procedures are given, and we show how a GaN(000 $\bar{1}$) surface was prepared for this study. Then results of diffraction and mass-spectroscopic measurements are presented, which show two growth regimes with different surface dynamics. These growth regimes are then explored in subsequent sections: the details of the diffraction measurements, the nature of the surface processes,

and the limits of growth under excess Ga and at high substrate temperatures. From these measurements we develop a simple kinetic model and framework that describes the overall growth.

2. Experimental

2.1. Instrumentation

Growth was carried out in a cryopumped Varian GEN II MBE system, which is shown schematically in Fig. 1. A quadrupole mass spectrometer mounted in one of the source ports enabled detection of the type and intensity of desorbed species from the sample surface. The specular RHEED intensity was monitored using a conventional RHEED system. Elemental Ga was provided by a Knudsen cell. The NH_3 flux was held constant during growth using a capacitance manometer in conjunction with a closed loop PID controller and solenoid control valve to maintain constant pressure in the NH_3 gas line behind a manually regulated precision leak valve. A Bayard-Alpert ionization gauge located on the back of the sample manipulator was used to determine the incident beam equivalent pressure (BEP) by rotating the sample manipulator. Absolute calibration of the incident Ga and Al fluxes was achieved by monitoring RHEED intensity oscillations during the growth of GaAs(001) and AlAs(001) in a separate experiment.

The sapphire substrate was heated by radiation from a high temperature resistive heater

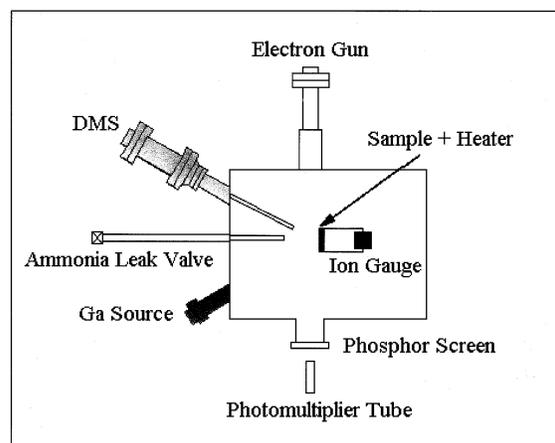


Fig. 1. Schematic diagram of the Gen II MBE system used for growth.

made from graphite loops deposited between pyrolytic boron nitride. A thermocouple touched the back of the heater, introducing an offset between the sample surface and thermocouple reading. This offset varied between 50° and 150°, depending upon the growth temperature, as well as from run to run, depending on the details of the sample mounting arrangement. This offset was compensated for by determining the surface temperature versus thermocouple reading, as described in detail elsewhere¹¹ and described in App. B.

A 10 keV electron gun was employed for RHEED measurements of the specular intensity without energy filtering. Measurements were made with the electron beam directed along the GaN $\langle 11\bar{2}0 \rangle$ azimuth. The diffracted intensity of the specular beam was measured using a phosphor screen and a photomultiplier tube.

Our DMS apparatus is based on a differentially pumped UTI 100 C quadrupole mass spectrometer mounted on one of the source flanges of the MBE system, permitting detection of the type and intensity of desorbed species from the sample surface. A similar apparatus has been used by Tsao *et al.*¹⁶ to study GaAs growth kinetics, and by several workers^{6,11,14,17,18} for GaN growth. The solid angle seen by DMS is limited by the cryoshroud in the growth chamber to subtend only the region around the sample holder. For a few of the measurements to be presented, additional collimation was added in order to limit this region so that only the center of the sample contributed to the desorption signal. This additional collimation reduced the effect of a temperature gradient across the sample as well as the background contribution of surrounding cooler surfaces. Since only a small portion of the sample was subtended, the collimation also reduced the signal-to-noise ratio. Those measurements performed with the additional collimation are indicated.

2.2. Sample preparation

For our DMS and RHEED measurements we used polished *c*-plane Al₂O₃ substrates, both 1 in. in diameter and 10 × 10 mm squares. The substrates were cleaned for 5 min in acetone and methanol, followed by a 5 min etch in 1:1 H₃PO₄:H₂SO₄. Each step was performed at 65°C, followed by a final deionized water rinse and rapid drying with high purity N₂.

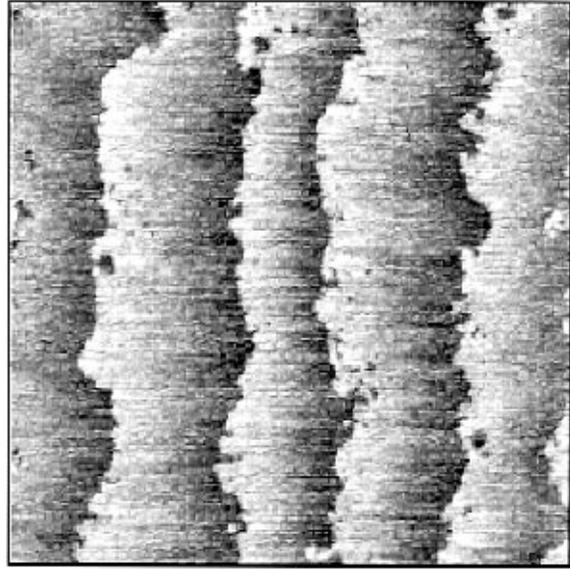


Fig. 2. A 500 × 500 nm AFM image of an atomically smooth Al₂O₃ substrate after being cleaned in acetone and methanol, followed by an etch in 1:1 H₃PO₄:H₂SO₄. Each step was performed at 65°C for 5 min, followed by a final deionized water rinse and rapid drying with high purity N₂. The steps are about 1000 Å in width.

Atomic force microscopy (AFM), as shown in Fig. 2, confirmed atomically smooth substrates featuring atomic steps. A 2000 Å Ti coating was applied to the back of the substrates by e-beam evaporation in a separate chamber. The coating serves to increase the heat absorbed by radiation and to assist in even heat distribution.

Mounting arrangements involving In bonding were found to be unreliable due to evaporation at the high substrate temperatures used in AlN and GaN growth, resulting in cold spots and eventual loss of the sample. Clips that applied even slight pressure induced unwanted stress on the sample and caused sapphire to crack during thermal cycling. Instead we used mounts which hold the samples loosely without stress. For 1 in. diameter substrates, this was accomplished by using a retainer ring which screws onto a Mo block. The Mo block has a hole at the center, slightly less than 1 in., exposing the back of the substrate directly to the heater. Figure 3 shows the type of sample holder used for square substrates, which has the ability to accommodate any shape and size sample from 3 × 3 mm to 0.5 × 0.5 in. via notched interleaving sliders.

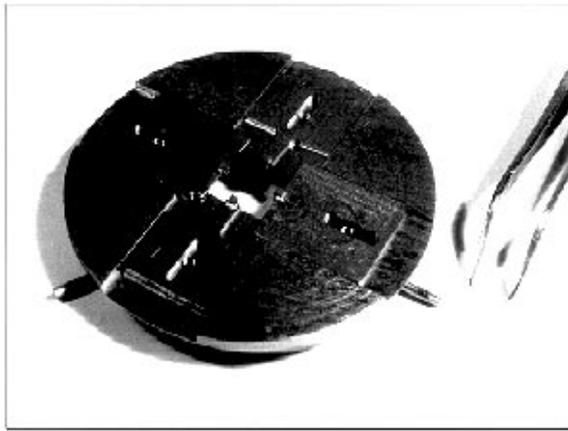


Fig. 3. Sample holder to accommodate samples stress free of any shape and size, from 3×3 mm to $1/2 \times 1/2''$. This is accomplished by four notched interleaving sliders mounted onto a standard Varian GEN II sample block.

2.3. Nucleation and buffer layer

Prior to growth the substrates were outgassed for several hours at 300°C in the preparation chamber of the MBE system, followed by 1 h at 500°C in the growth chamber. The NH_3 leak valve was set to produce a BEP of 1.0×10^{-5} Torr. Then the substrate temperature was ramped at $100^\circ/\text{min}$ from 500°C to 1000°C for a 15 min surface nitridation. AFM showed that the surface remained atomically smooth after this Al_2O_3 nitridation.

Experiments were performed by using two types of nucleation layer, AlN and GaN. The 250 \AA AlN nucleation layer was grown at 1000°C using an Al flux of $1.2 \times 10^{14} \text{ cm}^{-2} \text{ s}^{-1}$ (equivalent to 0.10 ML/s AlN). RHEED showed a transmission pattern during and after AlN growth. The Ga shutter was then opened to provide a flux of $1.1 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$ (or 1.0 ML/s GaN) and the sample temperature was ramped down to 800°C at $100^\circ/\text{min}$. The Al shutter was closed during the ramp below 900°C .

The GaN nucleation layer was grown by repeating a sequence consisting of condensing Ga on the substrate surface at 700°C and subsequent annealing at 800°C under NH_3 ten times. This was achieved by ramping the substrate temperature to 700°C , in the presence of a NH_3 BEP of 1.0×10^{-5} Torr. The Ga shutter was then opened at $1.1 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$ (or 1.0 ML/s GaN) and the substrate kept at 700°C for

about 1 min, during which time the RHEED pattern completely disappeared. The Ga shutter was then closed and the sample ramped to 800°C , where it was kept for about 2 min until a RHEED pattern reappeared. This procedure was repeated ten times, and a transmission RHEED pattern was observed after the completion of this nucleation procedure.

Both types of nucleation layers were followed by a 2500 \AA GaN buffer layer, deposited at 800°C with a Ga flux of $1.1 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$ (or 1.0 ML/s GaN) and an NH_3 BEP of 1.0×10^{-5} Torr. After the completion of the buffer layer, RHEED showed a streaky pattern. A 1×1 pattern was observed in the case of using GaN nucleation, while a very weak 2×2 reconstruction was typically observed using AlN nucleation. For this study, films grown with an AlN nucleation layer were found to be predominantly GaN(000 $\bar{1}$)-oriented, featuring inversion domains, while the GaN initiation layer resulted in unipolar GaN(000 $\bar{1}$) surfaces.⁸

3. Results

3.1. Overview

In this review we will describe different growth kinetics and growth morphologies obtained as a function of deposition parameters. We should emphasize, however, that the results depend on which of the two basal plane surfaces, either (0001) or (000 $\bar{1}$), is used for growth. These have different polarities and bonding configurations. A growth surface could also have inversion domains and be a mix of the two polarities. We use the usual III–V convention, that the [0001] direction is from a Ga to a N atom. Hence, if a GaN(000 $\bar{1}$) surface were terminated with a layer of N atoms according to the bulk Wurtzite structure, each N atom would be bonded to three Ga atoms in the underlying layer. In this paper we will discuss growth on GaN(000 $\bar{1}$) only. We will show that for this polarity the termination can be modified by adsorption and that the main growth regimes correspond to conditions of either excess N or excess Ga.

The two main surface morphologies observed^a in the MBE growth of GaN(000 $\bar{1}$) on sapphire are illustrated by RHEED and AFM data in Fig. 4. The two films were grown under similar conditions but with

^aNote added in proof: Films grown under excess Ga and excess NH_3 both exhibit three-dimensional features. These features are flat-topped under excess Ga and strongly pointed under excess NH_3 . These result in the 2-D and 3-D diffraction patterns observed.

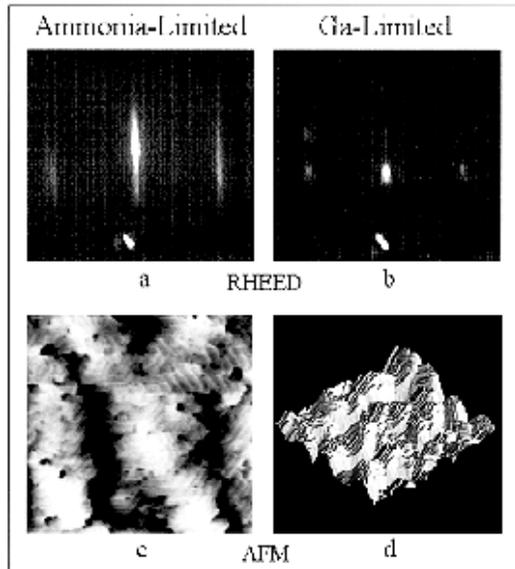


Fig. 4. RHEED patterns with the incident beam along the $\langle 11\bar{2}0 \rangle$ azimuth and AFM images for both NH₃- and Ga-limited growth. The AFM scans cover an area of $2 \times 2 \mu\text{m}$ with a z range of 50 Å and 1000 Å, respectively. NH₃-limited growth gives rise to a streaky RHEED pattern (a) and AFM shows atomic steps on islands (c). Ga-limited growth results in a RHEED pattern with a strong transmission component (b) and AFM indicates that this is due to facets (d). Note that for high Ga:NH₃ ratios pinholes with diameters of approximately 1000 Å are observed.

very different Ga fluxes. Both GaN films are 0.8 μm thick and were grown using an AlN nucleation layer, which typically produces some inversion domains.⁸ Growth rates were 0.5 $\mu\text{m}/\text{h}$ with a NH₃ BEP of 1.0×10^{-5} Torr at a substrate temperature of 800°C. Figures 4(a) and 4(c) show RHEED and AFM data for a Ga flux of $1.1 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$ (or 1.0 ML/s GaN) while Figs. 4(b) and 4(d) show data for a sample grown with a Ga flux of $6 \times 10^{14} \text{ cm}^{-2} \text{ s}^{-1}$ (or 0.5 ML/s GaN). AFM reveals that the film grown with a higher Ga flux is locally smoother, featuring atomic steps with 50 nm terraces, than the one grown with a lower Ga flux. Similar to what is shown by the AFM data, the RHEED pattern shows a well-defined specular beam with long streaks, indicative of two-dimensional layers with random atomic steps. The surface exhibited a very weak 2×2 reconstruction. By contrast, the film grown at a lower Ga flux is much rougher and faceted. Consistent with these AFM results, the RHEED pattern shows transmission-like features that do not move as the

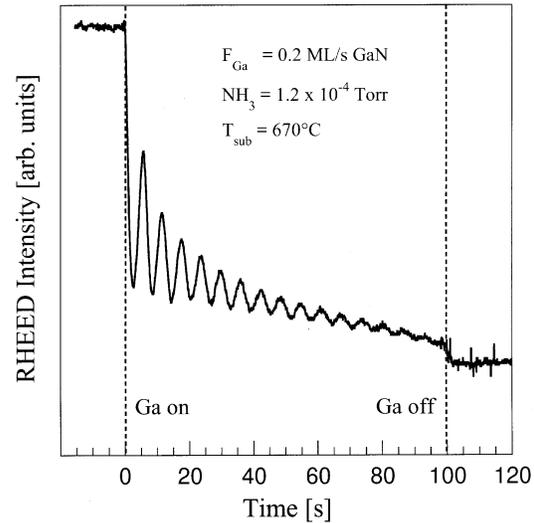


Fig. 5. GaN monolayer ($c/2$) RHEED intensity oscillations observed by first growing in the NH₃-limited regime until the RHEED pattern was streaky. Then the Ga shutter was closed, the NH₃ flux increased to provide Ga-limited conditions, and the substrate temperature lowered from 800°C to 670°C. Upon opening of the Ga shutter RHEED intensity oscillations were observable, recorded along the $\langle 11\bar{2}0 \rangle$ azimuth. Prolonged growth resulted in a 3D RHEED pattern. The film is best recovered by growth in the NH₃-limited regime.

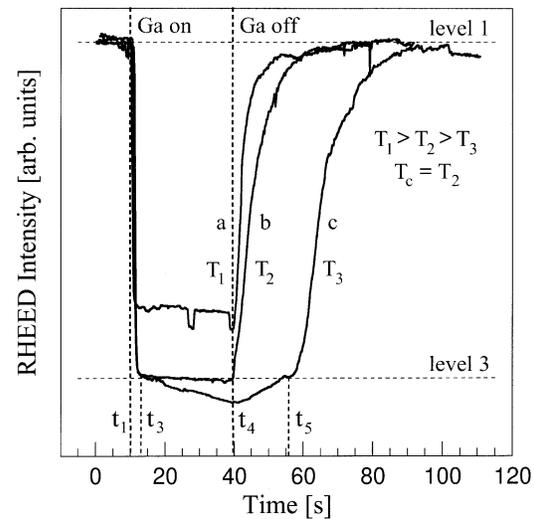


Fig. 6. Intensity variation of the specular RHEED beam along the $\langle 11\bar{2}0 \rangle$ azimuth during opening and closing of the Ga shutter in an NH₃ flux for (a) above, (b) at, and (c) below the condensation temperature. Curve (a) corresponds to a partial layer of weakly adsorbed Ga and partial nitridation, curve (b) to a complete layer of weakly adsorbed Ga, and curve (c) to Ga condensation.

incident electron angle is changed. Similar results are obtained on films with single domain GaN, grown with GaN nucleation layers, and on bulk GaN(000 $\bar{1}$), also free of inversion domains. These two morphologies are seen, in the following subsections, to correspond to distinct growth regimes.

The growth dynamics are also very different under these two Ga:NH₃ flux ratios. Two examples of the evolution of the surface structure, as measured by the specular RHEED intensity versus time, are shown in Figs. 5 and 6. In Fig. 5 we plot the intensity of the specular RHEED beam versus time for a sample held at 670°C for a low Ga:NH₃ flux ratio, while Fig. 6 shows a similar plot for a sample held at 780°C at a high Ga:NH₃ flux ratio. These are distinctly different behaviors, with one showing island nucleation and the other suggesting step flow. With DMS we will explore the rate-limiting reactions as well as the composition and reactivity of the surfaces corresponding to these growth conditions. We will show that island nucleation observed in these RHEED data corresponds to Ga-limited growth while step flow corresponds to NH₃-limited growth.

3.2. Growth regimes

To understand these distinct types of surface morphologies and growth kinetics, we used DMS to determine the rate-limited reactions. The main advantage of this technique is that measurements are made on the same sample under controllable growth conditions. We make the basic assumptions that the incident Ga either desorbs or contributes to GaN formation, and that no Ga inclusions or droplets are formed. Then, the difference between the incident and the desorbing Ga flux, as measured by DMS, is the growth rate. This method¹¹ does not require measurement of the sample area. It only requires a comparison of the desorbing Ga flux with and without an NH₃ flux. In conjunction with a calibrated Ga source, absolute growth rates are obtained. Figure 7(a) shows a measurement of the desorbing Ga flux with and without NH₃; the method of extracting growth rates from this data is discussed in more detail in App. A. This method works best when the desorbed Ga signal is large. With this method, we can measure absolute growth rates versus incident NH₃ and Ga flux as well as substrate temperature.

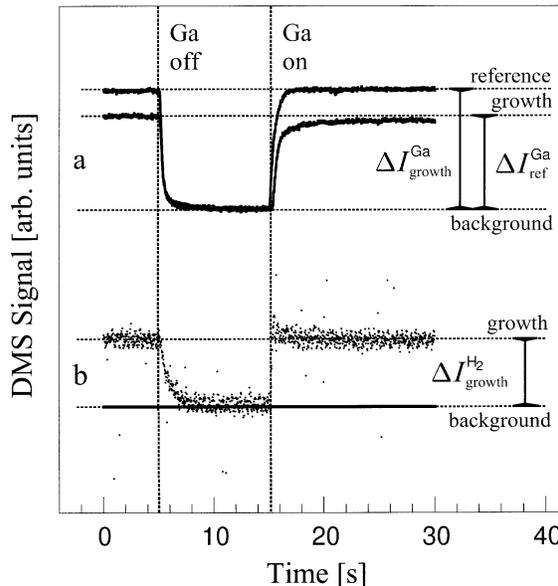


Fig. 7. Typical uncollimated Ga and H₂ DMS signal during closing and opening of the Ga shutter. The growth and reference signal is obtained by measuring the rapid change in signal after closing the Ga shutter to avoid contributions of the slowly varying background signal from other, cooler parts of the system. Growth rates are obtained from the Ga signals by calculating the incorporation fraction and multiplying by the known incident flux. The H₂ signal was found to be proportional to the growth rate. Note that the growth rate is obtained by closing the Ga shutter instead of opening to ensure steady state after 15 min of growth.

If the desorbed Ga flux is small, we instead use the desorbed H₂ signal. Figure 7(b) shows that H₂ is released during growth using NH₃. This figure is a plot of the H₂ signal versus time with the sample under constant NH₃ flux, while opening and closing the Ga shutter. Under the assumption that this increased H₂ desorption is attributed to the forward reaction



we can obtain relative growth rates. Measurement of the H₂ DMS signal is especially useful when the Ga signal is below the detection limit of the DMS. The H₂ signal intensity is larger because lower mass ions have a larger transmission through the mass spectrometer. A disadvantage of this technique is that absolute growth rates cannot be determined as easily. Nonetheless, in combination with measurement of the Ga signal, growth rates can be determined

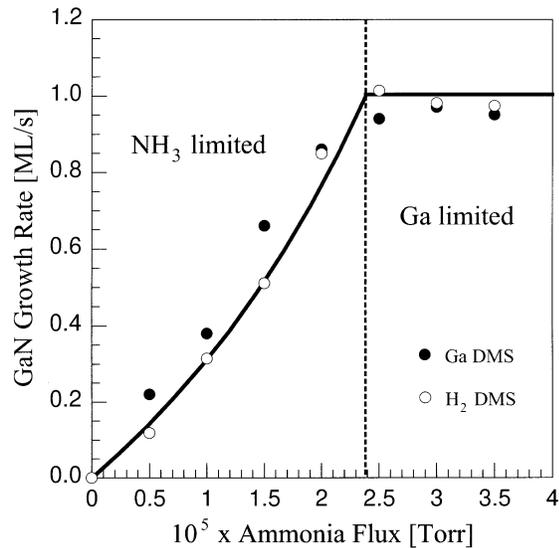


Fig. 8. GaN growth rate as a function of NH₃ flux at constant Ga flux of 1.0 ML/s (GaN) and surface temperature of 785°C, using both the Ga and the H₂ signal. Both growth regimes can be identified — NH₃-limited and Ga-limited. In the NH₃-limited regime the growth rate is an approximately linear function of the NH₃ flux, while in the Ga-limited regime near unity Ga incorporation is obtained. Note the good agreement between using Ga and H₂ DMS.

over a wide range of conditions. These DMS measurements that identify the rate-limiting fluxes are particularly important since RHEED is not able to measure growth rates under conditions of step flow growth.

Applying the above methods, Fig. 8 shows the GaN growth rate versus NH₃ flux at a fixed Ga flux and substrate temperature. These data represent a series of *in situ* measurements on the same sample, with 15 min of growth between data points under conditions of the point to be measured. The H₂ signal is proportional to the Ga data, indicating that using the H₂ signal is a valid method of obtaining relative growth rates. Growth rates calculated from the Ga signal agreed well with stylus and RBS measurements. From this figure a number of observations can be made. At low NH₃ fluxes the growth rate is approximately a linear function of the NH₃ flux, suggesting that a fixed fraction of the available NH₃ contributes to GaN formation. The deviation from linearity will be discussed later in terms of a proposed growth model. At high NH₃ fluxes no Ga desorption is measured. Hence the growth rate is a

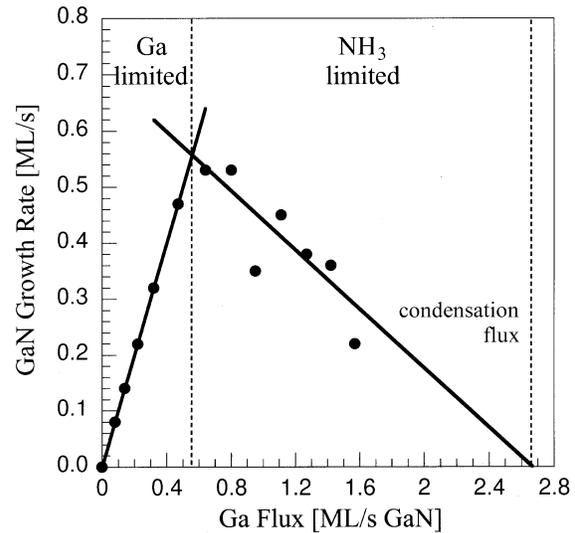


Fig. 9. GaN growth rate as a function of Ga flux at constant NH₃ flux of 1.0×10^{-5} Torr and surface temperature of 785°C, using Ga DMS. Two growth regimes can be identified — Ga-limited and NH₃-limited. Near unity incorporation of Ga is obtained during Ga-limited growth. During NH₃-limited growth the growth rate reduces as a function of Ga flux, approaching zero at the flux required for condensation at this temperature. This reduction is attributed to weakly adsorbed Ga blocking active Ga-limited GaN sites.

constant with unity Ga incorporation, within an uncertainty of less than 15%. The crossover between those two regimes is relatively abrupt, allowing the identification of at least two growth regimes: when there is excess Ga, the growth rate is limited by the NH₃ flux; when there is excess NH₃, the rate is limited by the Ga flux.

The crossover can also be seen by measuring the growth rate as a function of Ga flux, at a fixed NH₃ flux and substrate temperature. The results are shown in Fig. 9. From this figure we can make the following conclusions about the MBE growth of GaN(000 $\bar{1}$). At low Ga fluxes the growth is Ga-limited, as was observed above. At higher Ga fluxes a desorbed Ga signal was measured that indicates that not all the Ga was incorporated and that there is insufficient NH₃. At these fluxes growth is thus NH₃-limited. Further, the growth rate decreased with increasing Ga flux, going towards zero. In Sec. 3.5.1 we will show that the flux at which the growth ceased corresponds to the condensation condition for this substrate temperature. This can also be seen in

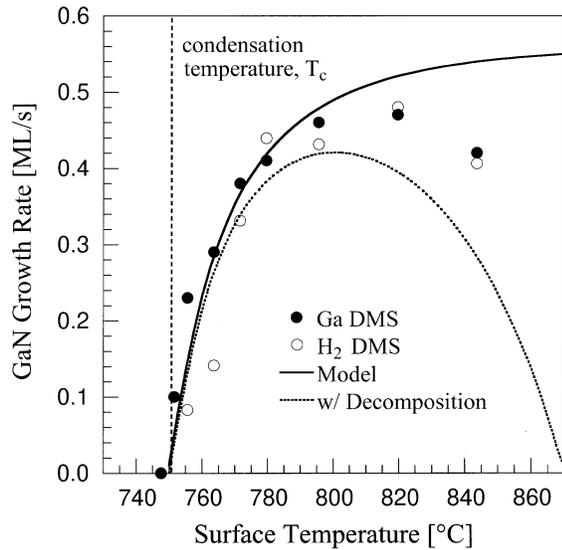


Fig. 10. GaN growth rate as a function of surface temperature at constant Ga flux of 1.0 ML/s (GaN) and NH_3 flux of 1.0×10^{-5} Torr, using both the Ga and the H_2 signal. The growth rate reduces towards the condensation temperature, attributed to weakly adsorbed Ga blocking active Ga-terminated GaN sites. The high temperature growth rate reduction is due to GaN decomposition competing with growth.

Fig. 10, where the growth rate is plotted as a function of substrate temperature in the NH_3 -limited regime.

3.3. RHEED measurements

3.3.1. Ga-limited growth

In the Ga-limited growth regime RHEED intensity oscillations were observed which indicate that epitaxy proceeds via the nucleation and growth of two-dimensional islands. An example of a measurement was shown in Fig. 5. The strongest oscillations were recorded below 700°C , but can be seen at temperatures as high as 850°C . Though intensity oscillations have been reported previously,^{19,20} these are, to our knowledge, the strongest observed to date. The period of these intensity oscillations, however, corresponds only approximately to monolayer growth.

Further, these intensity oscillations are not characteristic of a narrow growth front. Instead, continued growth eventually led to a transmission RHEED pattern. And, except perhaps at very high temperatures, the intensity does not recover upon closing the Ga shutter. In the case of other III-V semi-

conductor growth, such as GaAs(100), growth interruption at the growth temperature is enough to prepare a surface sufficiently smooth to subsequently observe intensity oscillations. But here, in order to repeat a series of intensity oscillations, an anneal to as high as 800°C was necessary. Alternatively a smoothing procedure involving growth in the NH_3 -limited regime was helpful. Growth in the NH_3 -limited regime, as described below, usually resulted in subsequently stronger oscillations than just annealing.

The process for smoothing by overgrowth starts by depositing approximately 100 \AA of GaN in the NH_3 -limited regime. Typical conditions are a substrate temperature of 800°C , a Ga flux of $1.1 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$ (or 1.0 ML/s GaN), and a NH_3 BEP of 1.0×10^{-5} Torr. After this smoothing step strong oscillations could again be observed. It should be noted that we observed these strong oscillations only when using a GaN nucleation layer, while in the case of an AlN nucleation layer oscillations were not observed or at best were exceedingly weak.

We also obtained damped AlN RHEED intensity oscillations by growing on GaN, as shown in Fig. 11, using the same conditions and surface preparation

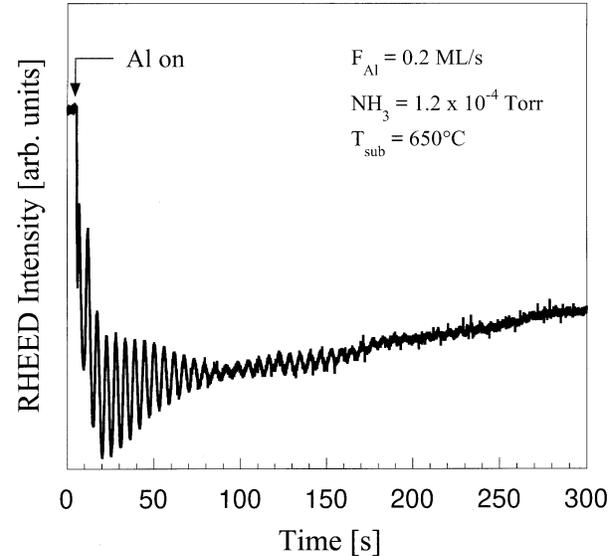


Fig. 11. AlN RHEED intensity oscillations were observed by preparing a GaN surface in the NH_3 -limited regime the same as for GaN oscillations. So far we were not able to smooth an AlN surface sufficiently to obtain AlN oscillation from an initial AlN surface. To our knowledge this is the first report of AlN oscillations.

procedures as for GaN on GaN. These oscillations correlate roughly with monolayer growth with unity Al incorporation. The GaN film was prepared by the previously described overgrowth method in the NH₃-limited regime. Then the substrate temperature was lowered to 670°C and the Al shutter opened at a flux of approximately 0.2 ML/s. The main requirement was preparation of a sufficiently smooth surface and growth under high NH₃ flux.

GaN RHEED intensity oscillations cannot be used directly to measure growth rate, as shown in Fig. 12. Here the frequency from data like those measured in Fig. 5 are plotted as a function of Ga flux at two different NH₃ fluxes. Contrary to previous suggestions,^{20,19} the period of these oscillations does not correspond exactly to monolayer (half a conventional unit cell) layer deposition. Note that the Ga flux was calibrated from GaAs intensity oscillations measured in the same growth chamber. The slope of the curve is about 0.9 for the fluxes investigated, and there is an offset of up to -0.08 ML/s at zero Ga flux. This offset depends on the NH₃ flux, increasing with increasing NH₃ flux.

The temperature dependence of the oscillations at different Ga fluxes is shown in Fig. 13. For these three fluxes no oscillations were observed outside the temperature range covered, and the strongest and

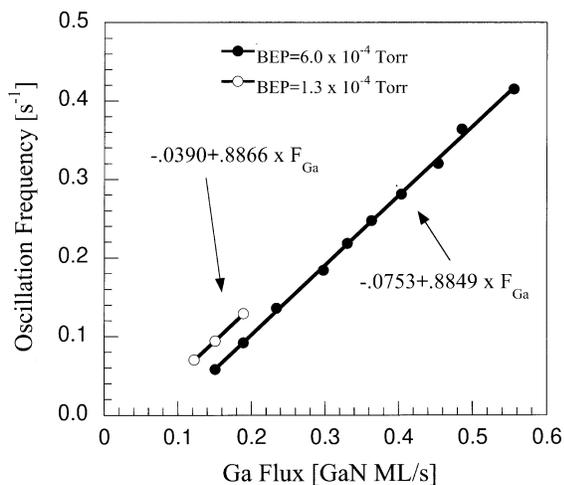


Fig. 12. GaN RHEED intensity oscillations at the low temperature limit show a linear increase in oscillation frequency as a function of Ga flux. The frequency does not quite correspond to monolayer growth. The slope is roughly 0.9, and an offset is observed. This offset increases with increasing NH₃ flux.

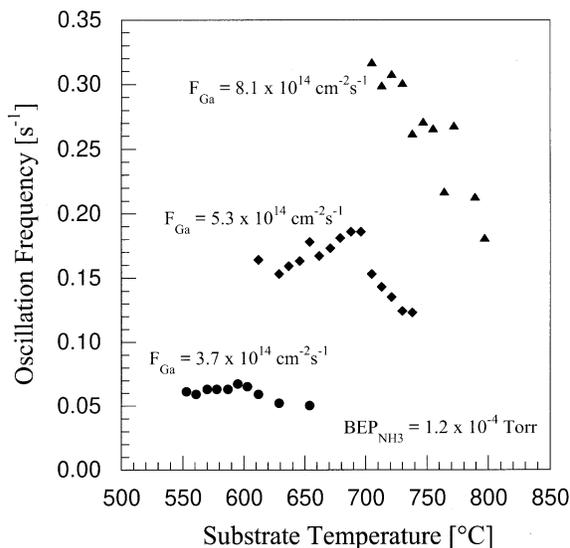


Fig. 13. GaN RHEED intensity oscillation frequency as a function of sample temperature for different Ga fluxes. The decrease in oscillation frequency at higher temperatures is attributed to a transition from island nucleation to step flow growth. The activation energy of this process was found to be 1.2 eV.

longest-lasting oscillations were observed at the lowest temperature of each curve. Below this temperature, we noticed that continued growth will not result in a transmission RHEED pattern, but a pattern that indicates amorphous growth, representing the low temperature limit of single crystal GaN growth in the Ga-limited regime. At high substrate temperatures the oscillation frequency decreases, and the activation energy of this decrease was found to be 1.2 eV. Other samples with miscuts ranging from 0 to $\pm 0.3^\circ$ off the *c* axis, including stress-free bulk GaN samples,⁸ gave similar results. This reduction is not solely due to Ga desorbing from the GaN surface, as verified by DMS earlier under similar conditions. We propose that this reduction in RHEED oscillation frequency is primarily due to Ga atoms diffusing to step edges, participating in step flow growth, and therefore not contributing to the oscillations.

Above about 850°C, with excess NH₃, intensity oscillations were not observed. At these temperatures, the streaks in the initial RHEED pattern from a smooth surface transform into a 3D transmission pattern of spots, and the corresponding RHEED intensity shows a slow decrease.

This transition to 3D was somewhat slower when the NH_3 flux was reduced. This is much different from the case of NH_3 -limited growth, showing a rapid initial decrease upon opening of the Ga shutter, as presented earlier in Fig. 6. The behavior at high temperature in the Ga-limited regime is related to the increase in the sublimation of GaN along with increased step flow.

3.3.2. NH_3 -limited growth

By contrast, for high Ga: NH_3 flux ratios, no intensity oscillations are observed, indicating that there is no cyclic change in surface roughness. This could still be a layer-by-layer mode. For example, the surface diffusion of Ga could be large, resulting in the step flow on the surface, as in the case of GaAs MBE when Ga stabilized. However, using these high Ga: NH_3 flux ratios, there is still a significant change in the diffracted intensity.

The key RHEED intensity behavior is shown in Fig. 6, where the specular RHEED intensity is plotted from a GaN surface during and after Ga exposure at three different substrate temperatures. The Ga shutter is opened at time t_1 and later closed at t_4 . The incident Ga and NH_3 fluxes were the same for all three curves. The starting RHEED intensity depends only on whether there has been sufficient initial NH_3 exposure. We used about 100 langmuirs, to completely nitride the surface as discussed in Sec. 3.4. (Note that 1 langmuir equals an exposure sufficient to adsorb 1 ML with unity incorporation.) This nitridation is relatively independent of typical growth temperatures. Three distinct behaviors are observed. Curve (a), measured at $T_s = 800^\circ\text{C}$, shows between t_1 and t_3 an initial rapid decrease to a value that is relatively constant, and then at time t_4 a rapid recovery. Curve (b), measured at a temperature 5°C lower, shows a similar but larger decrease to a constant value, as well as a rapid recovery. And curve (c), reduced by an additional 5°C , shows a similar initial behavior but then a slower steady decrease, followed by a two-stage recovery having a slow and a fast component. For the high and medium temperature cases, the magnitude of this decrease is a function of growth parameters. The initial decrease is less at higher substrate temperatures, higher NH_3 fluxes and lower Ga fluxes. For the low temperature case, the magnitude of the initial rapid decrease is

independent of temperature. However, the subsequent slower decrease becomes steeper with lower temperatures. After closing the Ga shutter the time required for the first stage of the two-stage recovery increases with lower temperatures.

The transition between curve (b) and curve (c) is very abrupt — a temperature change of a few degrees is sufficient for its observation. It was found that this transition is independent of NH_3 flux in the excess Ga regime (under excess NH_3 it is completely absent). We interpret the transition between curve (b) and curve (c) to correspond to the temperature, T_c , at which Ga condenses on the GaN surface. In Sec. 4.1 we will compare the RHEED data with DMS measurements, connecting the transition with the amount of adsorbed Ga.

3.4. Surface processes

3.4.1. Surface termination

Once again we emphasize that surface termination is different than crystal polarity. By “surface termination” we refer to the structure at the surface of a film which could be either GaN(0001) or GaN(000 $\bar{1}$) polarity. If the structure can be maintained without an incident flux at a given substrate temperature, it will be termed stable. We found that we can obtain and manipulate two stable surface terminations on GaN(000 $\bar{1}$). We used DMS and RHEED to control and monitor these terminations.

First a surface was prepared by exposure of a GaN(000 $\bar{1}$) film to a NH_3 flux in the absence of Ga. After a 100-langmuir exposure, the specular RHEED intensity had reached a maximum value. This nitrated surface is stable termination since it does not change after the NH_3 flux is removed and serves as the starting surface for the measurements in Fig. 14.

Preparation of a GaN(000 $\bar{1}$) surface with a Ga termination is illustrated in Fig. 14. A substrate temperature above that for Ga condensation was chosen (cf. Sec. 3.5.1). A nitrated surface was then exposed to a Ga flux in the absence of NH_3 while monitoring the specular diffracted intensity as well as the desorbing Ga. After opening the shutter at t_1 , the RHEED intensity decreased to a steady state value at t_3 , after going through a change in slope at t_2 . The time to reach this steady state depended on the Ga flux. Since this is a steady state condition,

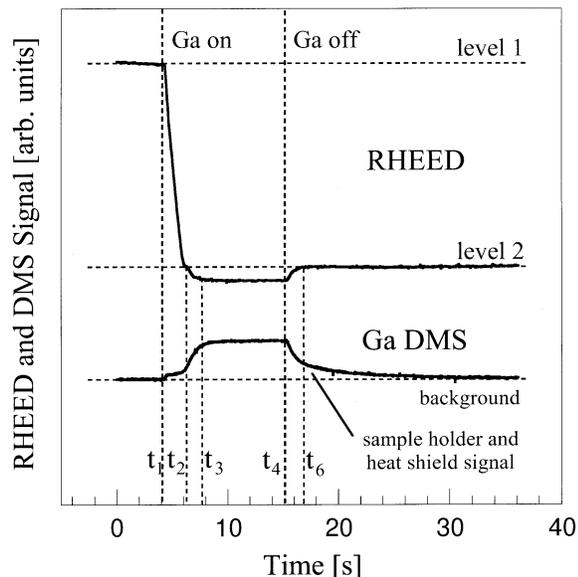


Fig. 14. RHEED and uncollimated Ga DMS signal without NH₃ flux above the condensation temperature. Level 1 indicates a nitrated starting surface, while a stable level 2 is reached after Ga exposure, indicating a Ga-terminated surface. Note that during Ga exposure the intensity level reaches steady state below level 2, suggesting fractional coverage of weakly adsorbed Ga.

the Ga desorption flux is equal to the incident flux. The Ga shutter was then closed at t_4 , at which time the RHEED intensity increased until it reached at t_6 an intensity marked level 2, and the DMS signal simultaneously decreased to a background. The final RHEED intensity was independent of the steady state Ga coverage that was achieved during deposition, when the RHEED intensity was below level 2. The DMS shows Ga desorbing, beyond t_4 , which must be weakly bound compared to any Ga remaining at t_6 . The last surface is stable in the absence of a Ga or NH₃ flux. Our interpretation of this data is that this surface consists of a strongly bound Ga termination layer on GaN(000 $\bar{1}$).

If the surface is subsequently reacted with NH₃, the behavior seen in Fig. 6 will again be obtained and the RHEED intensity rises to the nitrated value of level 1. These observations indicate that there are at least two stable surface terminations possible on GaN(000 $\bar{1}$) — one is reached after the GaN surface is exposed to an NH₃ flux in the absence of Ga (level 1), and the other is obtained after Ga exposure, followed by an anneal in the absence of both NH₃ and Ga (level 2).

Knowing how to prepare these two surface terminations on GaN(000 $\bar{1}$), we looked for the reconstructions reported by Smith *et al.*,⁹ who used a hydrogen-free, RF plasma as a nitrogen source. We found that in order to obtain the same reconstructions, the Ga termination had to be prepared first. Then the sample temperature was lowered to 400°C. After deposition of additional Ga, the substrate temperature was lowered to below 300°C and all of the reconstructions reported by Smith were observed, depending on the coverage of weakly bound Ga deposited.

3.4.2. Surface reactivity

We now investigate the reactivity of Ga on a previously nitrated surface, which will give us a better understanding of the growth process in the NH₃-limited regime.

The transient response of the specular RHEED intensity as well as the response of the desorbed Ga and H₂ fluxes to a step function of incident Ga on the GaN surface are shown in Fig. 15. These data were measured after growth had been stopped and the surface was nitrated. Then the background NH₃ pressure was reduced to 10⁻⁹ Torr while maintaining a constant substrate temperature of 760°C.

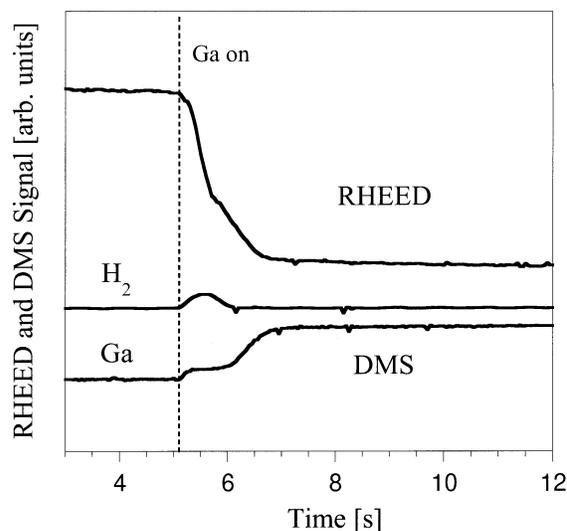


Fig. 15. The transient response of the Ga and H₂ desorption to a step function of incident Ga is shown in the absence of incident NH₃. The response of the specular RHEED intensity observed along the $\langle 11\bar{2}0 \rangle$ azimuth is also shown. ($F_{\text{Ga}} = 1.6 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$, $T_{\text{sub}} = 760^\circ\text{C}$.)

A number of important features are observed in Fig. 15 after opening the Ga shutter at a substrate temperature above Ga condensation. The initial shoulder in the curve corresponds to an initially low Ga desorption flux. This indicates that Ga adsorbs in a strongly bound site until the available sites are saturated. We also see a H_2 pulse that is a by-product of this adsorption process, indicating that this surface contains H.¹⁴ After deposition of roughly 1 ML, the Ga desorption increases and the H_2 desorption decreases, indicating the presence of a second, weakly bound state. These results are consistent with the data showing two adsorption sites presented in Sec. 3.4.1. After the initial Ga pulse, we close the Ga shutter and allow the weakly bound Ga to desorb from the surface. Subsequent exposure to incident Ga results in only the higher Ga desorption flux, with no detectable change in the H_2 desorption. This indicates that the hydrogen left in the nitridation process has been depleted. The hydrogen released required approximately 1 ML of incident Ga, deduced from the area under the desorbed Ga curve. Further, no additional Ga is adsorbed into the strongly bound sites, confirming that those sites are stable with time in the absence of any fluxes. Subsequently we could again prepare a surface that would adsorb Ga in the strongly bound sites by exposing the sample to NH_3 .

To quantify the Ga adsorption we measured the RHEED intensity and plotted its derivative for different growth conditions. Figure 16 shows a measurement which was performed under NH_3 exposure while opening the Ga shutter. The derivative clearly reveals that two slope maxima occur during the initial transient RHEED decrease. Quantitative information can be extracted from this RHEED intensity variation by definition of a time interval, Δt , as indicated in Fig. 16. We find that the time dependence of the transient Ga and H_2 desorption tracks the variations in the RHEED intensity. Based on this observation we conclude that Δt gives an estimate of the time required for saturation of the strongly bound Ga sites.

The dependence of Δt on both NH_3 and Ga flux is shown in Fig. 17, which shows $1/\Delta t$ versus NH_3 BEP for three different Ga fluxes. We see that $1/\Delta t$ increases for increasing Ga flux, but that it decreases as the incident NH_3 flux is increased. We note that the intercept on the $1/\Delta t$ axis is nearly equal to the

incident Ga flux for all three curves, again suggesting that 1 ML of Ga can be adsorbed in strongly bound sites on a nitrided surface. The decrease in $1/\Delta t$ with increasing NH_3 pressure indicates that the physical process responsible for these transients is slowed by increasing the amount of incident NH_3 .

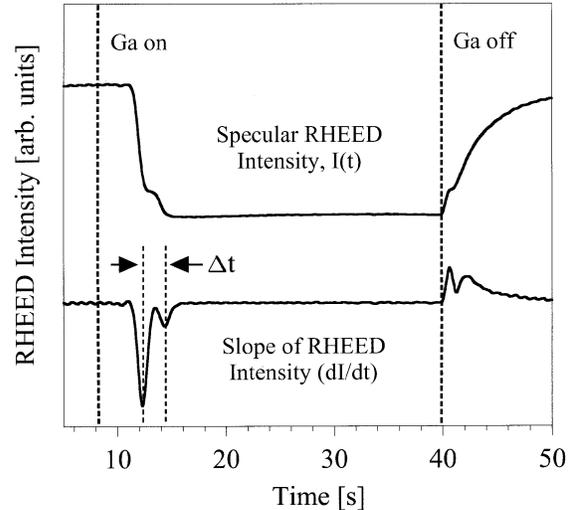


Fig. 16. Specular RHEED intensity and its first derivative. Differentiation of the signal allows quantitative analysis of the transient signal. Experimental conditions for the data shown here are $T_{\text{sub}} = 780^\circ\text{C}$, NH_3 BEP = 1.1×10^{-5} Torr, $F_{\text{Ga}} = 1.45$ ML/s.

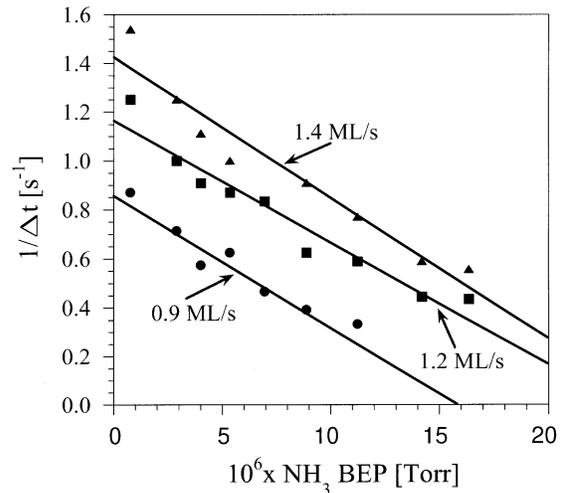


Fig. 17. Dependence of the surface transformation rate, $1/\Delta t$, on incident NH_3 BEP, at different Ga fluxes. The substrate temperature is 780°C . The linear curve fits shown in the figure were obtained using the least squares method.

A simple first order estimate of the time required for a termination change from a nitrated surface to a surface with strongly bound Ga sites can account for this behavior. In this model we assume that the Ga diffusion length is large so that every incident Ga atom will react with the nitrated surface. We want to estimate the time required for the entire surface to become Ga-terminated. We neglect the small desorption of Ga from the strongly bound sites observed in Fig. 15. The time derivative of the N-H surface coverage, θ_N , is given by the difference between the incident fluxes

$$\frac{d\theta_N}{dt} = F_{\text{Ga}} - kP_{\text{NH}_3}, \quad (3.2)$$

where F_{Ga} is the incident Ga flux in ML/s, and kP_{NH_3} is the active N flux in ML/s as supplied by the NH₃, with k being the system-dependent conversion factor between measured NH₃ BEP and active N flux. Solving Eq. (3.2) subject to the initial condition $\theta_N = 1$ at $t = 0$, we find that

$$\theta_N = 1 - (kP_{\text{NH}_3} - F_{\text{Ga}})t, \quad (3.3)$$

and from this we see that $\theta_N = 0$ when $t = 1/(F_{\text{Ga}} - kP_{\text{NH}_3})$, which is the time required to change all of the surface sites from N(H)-terminated to Ga-terminated. Consistent with this simple picture of the surface kinetics is the observation that the linear curve fits shown in Fig. 17 intercept with the $1/\Delta t$ axis close to the incident Ga flux, as expected from Eq. (3.3).

3.4.3. Transient behavior

In this section we investigate the transient DMS response to a pulse of Ga under conditions of excess N and excess Ga. Figures 18 and 19 show the desorbing H₂ and Ga fluxes resulting from exposure of the GaN surface to a 15 s pulse of Ga while the NH₃ BEP is held constant. For the data shown in Fig. 18, representing the Ga-limited regime, the Ga flux was $4.2 \times 10^{14} \text{ cm}^{-2} \text{ s}^{-1}$. In Fig. 19, representing the NH₃-limited regime, it was increased to $1.4 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$. All other growth parameters were the same for the two plots. In Fig. 19 we see that initiation of growth causes a transient pulse of H₂ to desorb from the surface, whereas in Fig. 18 the desorbed H₂ flux reaches its maximum value at steady state. A similar H₂ pulse is observed after

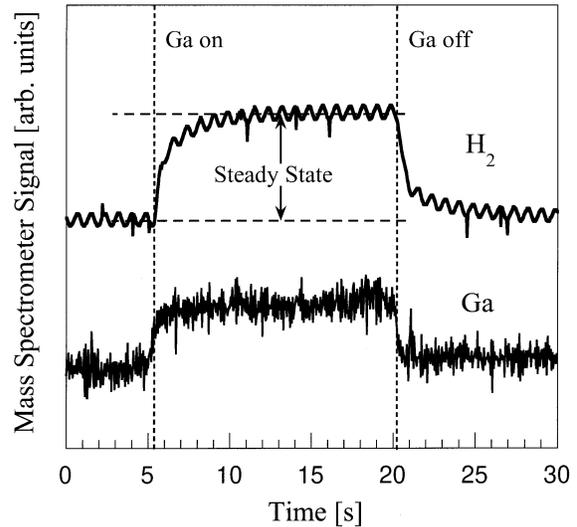


Fig. 18. Changes in the H₂ and Ga desorption are caused by exposing a smooth GaN sample to a 15 s pulse of incident Ga under excess NH₃ conditions. $T_{\text{sub}} = 820^\circ\text{C}$, NH_3 BEP = 7×10^{-6} Torr, $F_{\text{Ga}} = 4.2 \times 10^{14} \text{ cm}^{-2} \text{ s}^{-1}$. The high frequency H₂ signal oscillations arise from fluctuations in H₂ background pressure caused by temperature cycling of the cryopumps.

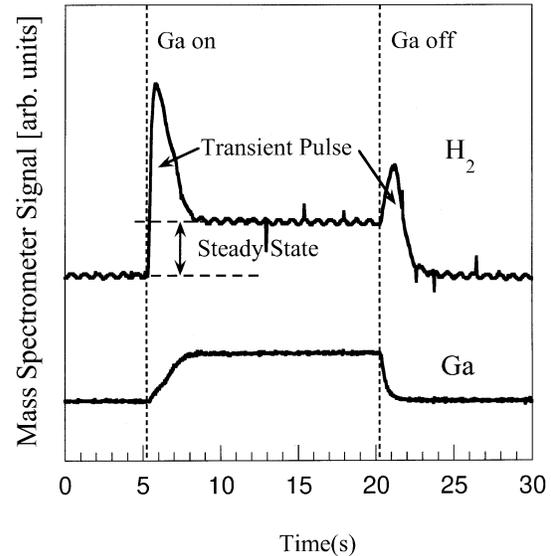


Fig. 19. Changes in the H₂ and Ga desorption are caused by exposing a smooth GaN sample to a 15 s pulse of incident Ga under excess Ga conditions. $T_{\text{sub}} = 820^\circ\text{C}$, NH_3 BEP = 7×10^{-6} Torr, $F_{\text{Ga}} = 1.4 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$. The H₂ signal oscillations arise from fluctuations in H₂ background pressure caused by temperature cycling of the cryopumps. Note the pulse in the H₂ signal after opening and closing the Ga shutter.

closing the Ga shutter in the NH_3 -limited regime, also seen in Fig. 19.

The data in Fig. 19 point to a crucial aspect of growth under conditions of excess Ga. If we take the H_2 signal to be proportional to the GaN growth rate, these data indicate that the growth rate is enhanced in this NH_3 -limited regime just after opening and closing the Ga shutter. We have shown in Secs. 3.4.1 and 3.4.2 that Ga adsorbs in two adsorption sites, strongly bound and weakly bound. After closing the Ga shutter the weakly adsorbed Ga either desorbs or is consumed by active N sites. Similarly, just after initiating the Ga flux, Ga first adsorbs into strongly bound sites, and then into weakly bound sites. The enhancement occurs during those times when the coverage of weakly adsorbed Ga is lower than that at steady state during growth. This behavior suggests that the presence of weakly bound Ga reduces the growth rate. In other words, this suggests that when excess Ga atoms are in the lattice sites deduced by Smith *et al.*⁹ the growth rate is reduced from its maximum value. We will include this in a first order model¹⁴ in Sec. 4.2.

3.5. Limits of growth

3.5.1. Ga condensation

One of the important limits of GaN growth is the weak adsorption of Ga on the GaN surface. This is relevant at low substrate temperature or high Ga flux. The adsorption behavior can be seen in Fig. 20, which shows the desorption behavior of Ga as a function of substrate temperature. Desorption is shown both with an NH_3 flux and without an NH_3 flux in the excess Ga regime. The substrate temperature was calibrated beforehand under the assumption that the condensation behavior is well described by liquid Ga vapor pressure data. This method of measurement is consistent with temperature measurements described elsewhere.¹¹ The key point of the figure is that there is a temperature at which the desorbed Ga flux changes abruptly. This behavior is seen with and without an NH_3 flux and is independent of the magnitude of the NH_3 flux in the excess Ga regime. The temperature depends only on the incident Ga flux. Just above this temperature the NH_3 reduces the Ga desorption, indicating that there is growth of GaN. Just below this temperature, Ga

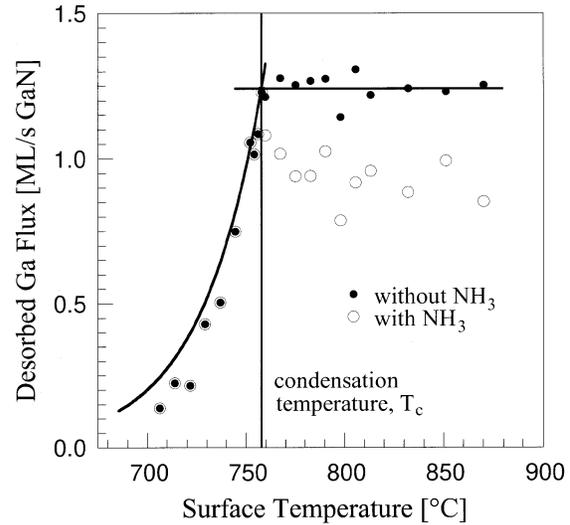


Fig. 20. Condensation temperature determination using collimated Ga DMS with and without NH_3 for an incident Ga flux of 1.25 ML/s GaN. Without NH_3 the DMS signal remains constant above the condensation temperature, while with NH_3 the signal is reduced due to GaN formation. The condensation temperature is found as illustrated by observing the temperature at which the signal drops exponentially. No growth is observed below the condensation temperature. The solid curve at temperatures below T_c is calculated from Ga vapor pressure data, closely followed by the experimental data. (The data presented in this figure should not be used for growth rate determination, since steady state was not obtained.)

desorption is independent of the NH_3 flux, indicating that there is no growth.²¹ This abrupt transition temperature is very reproducible and is part of the model discussed in Sec. 4.2.

This same adsorption behavior in this excess Ga regime was also seen in the RHEED data of Fig. 6, where we interpreted the transition between curve (b) and curve (c) to correspond to the temperature, T_c , at which Ga condenses. This behavior was not observed in the Ga-limited regime. Instead, RHEED intensity oscillations, as shown in Fig. 5, were recorded at low temperatures, their frequency indicating near unity Ga incorporation. These results indicate that the GaN growth kinetics in those two regimes differ significantly. We identify the abrupt transition temperature, between growth and no growth, as the temperature at which multilayers of Ga begin to condense on the surface.

To correlate RHEED and DMS measurements in terms of Ga condensation and surface termina-

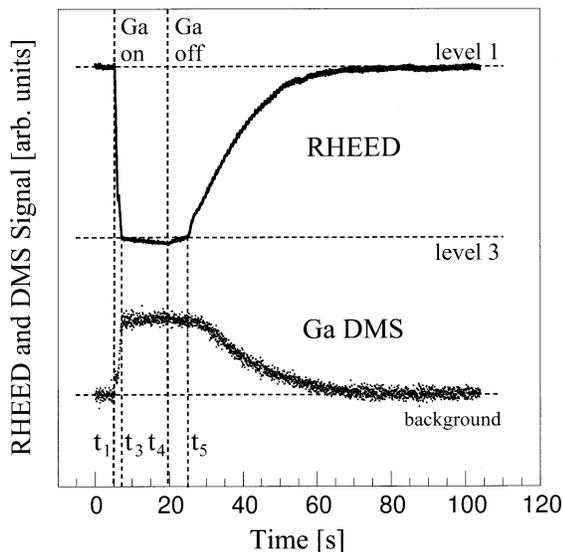


Fig. 21. RHEED and uncollimated Ga DMS signal during Ga accumulation below the condensation temperature. After the Ga shutter is closed at t_4 Ga desorption remains constant until part of the GaN surface is exposed at t_5 . At the same time, the RHEED intensity increases slowly up to t_5 , where a large change in slope takes place. The slow decrease of the DMS signal after t_5 is partly due to the temperature gradient across the sample, resulting in unsynchronized completion of Ga desorption from the surface.

tion, we compare in Fig. 21 measurements of the desorbed Ga to the corresponding RHEED intensity versus time, at a temperature below T_c at which Ga condenses on the surface. To improve the signal-to-noise ratio, we did not use collimation for this set of DMS measurements and therefore the DMS averages over the entire surface of the 1 in. sample. The RHEED measurement, on the other hand, is more localized, and by moving the beam across the surface we determined that there was an approximately 30° temperature gradient across the sample surface. After opening the Ga shutter at t_1 , the decrease in the RHEED intensity parallels an increase in the Ga DMS signal to a steady state value at t_3 . There is a slight decrease in the RHEED intensity after t_3 , due to increased attenuation of the signal originating from ordered regions. By contrast, the Ga DMS signal is constant after t_3 . This is consistent if the desorption energies from successive layers are approximately equal. The magnitude of this desorbed Ga is independent of temperature above T_c and then decreases below T_c due to condensation.

The recovery behaviors of the two signals after the Ga source is shuttered are also correlated to an extent — the DMS indicates that the increase in the diffracted intensity can be associated with a decrease in the amount of adsorbed Ga and not just to a coalescence of surface adatoms that reduces the step density. The correlation is not exact, however. After the Ga source is shuttered at t_4 the RHEED intensity increases slightly until, at t_5 , there is a rapid increase. The DMS signal remains constant during this time and then decreases to a background. A difference in recovery rates would be expected if regions on the sample surface at lower temperature, where there is a larger amount of condensed Ga to desorb, also contribute to the DMS signal. Thus the condensation is seen in both the RHEED and DMS data, though regions sampled can be quite different.

3.5.2. GaN decomposition

GaN decomposition must be considered during both NH₃-limited and Ga-limited growth at high substrate temperatures in order to be able to predict growth rates. In the following section we will review data on the decomposition of GaN, which represents the fundamental limit to its growth at high temperatures. Determination of the GaN decomposition rate was carried out after growth of about $1 \mu\text{m}$ of GaN, and after the base pressure of the system was reduced to the mid 10^{-9} Torr range. Residual gas analysis indicated that the gas remaining in the system was predominantly NH₃.

Determination of the GaN decomposition rate was accomplished by monitoring both Ga and molecular nitrogen signals while varying the substrate temperature. The Ga signal coming from the GaN sample, as detected by the DMS, was calibrated previously in terms of absolute flux. This was achieved by exposing the GaN sample with a known Ga flux above the condensation temperature, at which the incident flux equals the desorbing flux. The incident flux was known from the measurement of RHEED intensity oscillations during the growth of GaAs(001) in a separate experiment.

Quantification of the N₂ flux desorbing from the substrate was not possible by direct means. However, a useful comparison between N₂ and Ga signals can be obtained assuming that sublimation is congruent. In this case, the N₂ flux leaving the substrate is

related to the Ga flux leaving the substrate via the relation

$$F_{N_2} = \frac{1}{2}F_{Ga}. \quad (3.4)$$

Hence, to within a scale factor that depends on the sensitivity of the mass spectrometer, the two signals should be equal. In Fig. 22 we show a plot of $1/2 F_{N_2}$ and F_{Ga} versus substrate temperature resulting from GaN decomposition. The scale factor used to obtain the absolute N_2 flux is based on the assumption of congruent sublimation. The assumption is justified by the fact that the slopes of the two curves are the same to within the experimental error. We see that the data presented here are consistent with previous measurements,^{22,b} and also that these results extend the temperature range of available experimental data. The activation energy of GaN decomposition was therefore verified to be 3.4 eV in the experimental range covered. It was found that the GaN

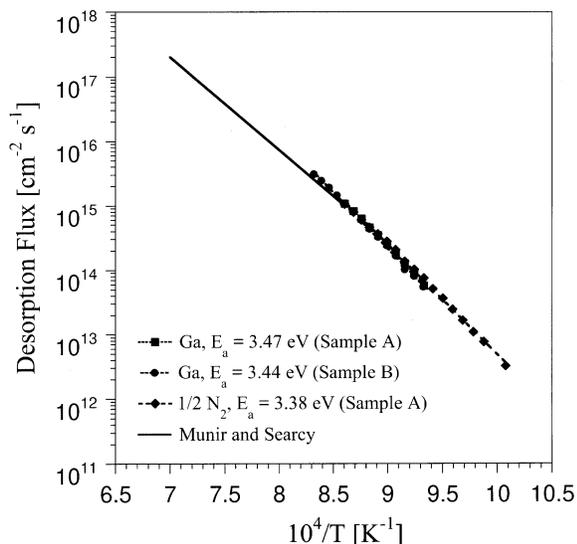


Fig. 22. The rate of GaN decomposition compared with values previously reported by Munir and Searcy. The data presented here extend the experimental curve to lower temperatures. The current measured on the mass spectrometer while monitoring desorption of molecular nitrogen was multiplied by a constant factor such that the signals are equal in the middle of the range of measured values. The close agreement between the activation energies of the $1/2 N_2$ curve and the Ga curve indicates that decomposition is congruent.

decomposition rate reduces in the presence of Ga, consistent with the law of mass action, published elsewhere in detail.²³ The exact dependence of GaN decomposition in the two growth regimes, as a function of both Ga and NH_3 fluxes, still has to be investigated and we expect the results to follow the law of mass action. The main point is that at growth temperatures above $800^\circ C$ the growth rate can be reduced significantly due to GaN decomposition, which cannot be neglected in an overall growth model.

4. Discussion

4.1. Ga adsorption

The experiments presented in Sec. 3.4 have identified two Ga adsorption sites on GaN(000 $\bar{1}$) which can be inferred from both RHEED and DMS data. We have identified strongly adsorbed Ga which does not desorb in the temperature range investigated, as well as weakly adsorbed Ga, on top of the strongly adsorbed Ga sites, which desorbs according to Ga vapor pressure data in the absence of any incident fluxes. The weakly adsorbed Ga is necessary for giving the low temperature reconstructions observed by Smith *et al.*⁹ Our results are consistent with the findings of Jones *et al.*⁶ and Lee *et al.*,²⁴ who both proposed that Ga exists in two adsorption sites during growth. At typical growth temperatures, this Ga-terminated GaN(000 $\bar{1}$) surface is stable in the absence of Ga and NH_3 .

The DMS and RHEED data presented in Sec. 3 can be understood by considering the adsorption of Ga onto an otherwise inert surface in the absence of NH_3 . Let us examine two limits. First, at sufficiently high temperature, an incident Ga flux will produce a steady state Ga coverage that depends on the incident Ga flux and the Ga residence time. In this steady state, the total amount of adsorbed Ga is constant so that the incident flux must equal the desorbing flux. Second, if the substrate temperature is decreased sufficiently, there will be a temperature below which Ga condenses on the surface, forming multilayers and/or droplets. Below this temperature the system is not in steady state since the incident flux exceeds the desorbing flux, and Ga is adsorbed continuously. The transition from steady state to

^bNote added in proof: Munir and Searcy examined polycrystalline samples. Our recent measurements on bulk GaN(000 $\bar{1}$) show a roughening during decomposition which might indicate that other faces have higher rates of decomposition.

condensation is sharp and can be determined by measuring the temperature at which the desorbed Ga flux begins to decrease. For a given incident Ga flux, this transition corresponds to a unique surface temperature well described by equilibrium vapor pressure data.

The picture that emerges is illustrated in Fig. 23, where weakly bound Ga adsorbed on Ga-terminated GaN(000 $\bar{1}$) is shown. In this situation, if there were no NH₃ or Ga flux present, the weakly bound Ga desorbs, leaving the strongly bound Ga to make up the Ga-terminated GaN(000 $\bar{1}$) surface. If now an NH₃ flux is provided, the strongly bound Ga of the

Ga-terminated surface would react, resulting in a nitrided surface. If then a Ga flux is provided to this nitrided surface, in the absence of NH₃, the Ga would react with the nitrided surface, causing hydrogen to desorb. After the nitrogen is consumed by strongly bound Ga, a surface coverage of weakly bound Ga will build up. If the substrate temperature is above the condensation temperature the weakly bound Ga, likely less than a monolayer, will reach steady state, as shown in Fig. 23(a). If the substrate temperature is below the condensation temperature for that incident Ga flux, Ga multilayers and/or droplets will form, illustrated in Fig. 23(c). The crossover between steady state coverage of weakly adsorbed Ga and condensation is shown in Fig. 23(b). Once the Ga flux is stopped, any weakly bound Ga will again desorb to eventually expose the strongly bound Ga-terminated surface.

During growth the steady state coverage of weakly adsorbed Ga is maintained by the incident Ga minus the desorbing Ga and Ga forming GaN. We will present a model in Sec. 4.2 in which the strongly bound Ga contributes to growth, whereas the weakly bound Ga acts to inhibit growth by blocking these strongly bound Ga sites. The effect of weakly adsorbed Ga on the growth rate can only be observed in the NH₃-limited regime, while in the Ga-limited regime near unity incorporation of Ga is obtained, as discussed later in Sec. 4.3.

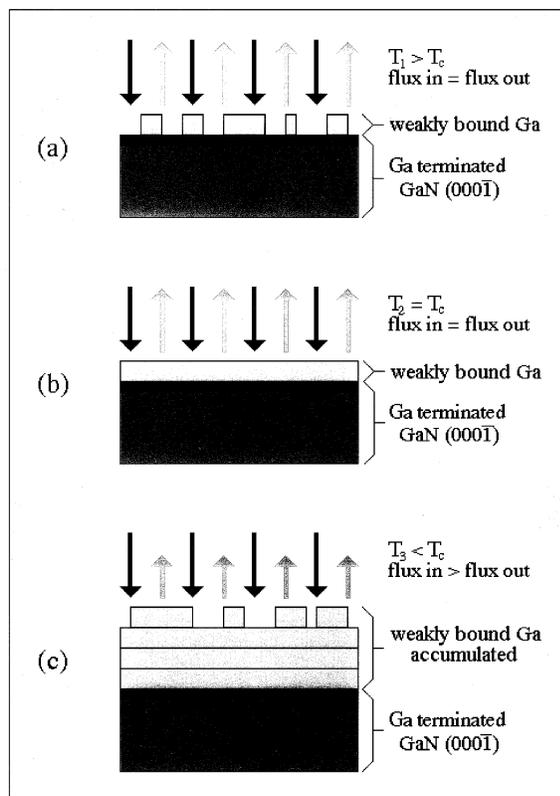


Fig. 23. Behavior of incident Ga on a rigidly attached (strongly bound) Ga termination layer as a function of temperature in the absence of NH₃. Above the condensation temperature (a) steady state partial Ga coverage is obtained, and the incident flux equals the desorbing flux. Complete steady state Ga coverage is obtained at a critical temperature (b) below which Ga condensation takes place. Note that the incident flux still equals the desorbing flux. Below this critical temperature (c), Ga multilayer accumulation takes place and the incident flux exceeds the desorbing flux.

4.2. NH₃-limited growth

As showed earlier in Fig. 9, at constant substrate temperature above Ga condensation and at constant NH₃ flux, the growth rate increases linearly with increasing Ga flux, but then decreases as the Ga flux exceeds a certain value. The reduced growth rate at high Ga fluxes can be accounted for by consideration of the following simple kinetic model. We start by making a number of assumptions about the atomistic behavior of adsorbed Ga which will be justified by agreement with measured data. First, we consider Ga in the weakly adsorbed state. The fractional area of the surface covered by weakly bound Ga is σ_{Ga} . The second assumption is that Ga desorption occurs only from this weakly bound site, which results in a Ga desorption term that is proportional to σ_{Ga} . We assume that for complete coverage ($\sigma_{\text{Ga}} = 1$) the desorption flux is equal to the evaporation rate of

Ga from liquid Ga, $F_0(T_{\text{sub}})$, and therefore approximates the Ga desorption rate as $\sigma_{\text{Ga}}F_0(T_{\text{sub}})$. The model does not require that Ga completely wet the GaN surface, since the shadowing effect of droplets could account for the observed behavior as well. In the limit of complete wetting, σ_{Ga} is the surface coverage of weakly adsorbed Ga.

We assume that NH_3 reacts only with strongly bound Ga, and that the excess Ga in the weakly bound site reduces the growth rate by blocking the underlying reactive Ga sites. Further motivation for such a growth mechanism can be found in the work of Liu and Stevenson,²⁵ who found that the coexistence of Ga and GaN enhanced the decomposition of NH_3 relative to Ga alone.

We let the growth rate be proportional to the fraction of strongly bound Ga sites that are exposed to the incident NH_3 , $(1 - \sigma_{\text{Ga}})$. DMS measurements of the H_2 flux showed that the NH_3 reactivity does not depend on substrate temperature over the range of 700°–820°C, and we therefore let the active N flux, F_{N} , be independent of substrate temperature. We wish to determine the steady state growth rate when the incident Ga flux exceeds the active N flux, $F_{\text{Ga}} > F_{\text{N}}$. The time derivative of the Ga coverage is given by

$$\frac{1}{A} \frac{dN_{\text{Ga}}}{dt} = F_{\text{Ga}} - F_{\text{N}}(1 - \sigma_{\text{Ga}}) - \sigma_{\text{Ga}}F_0(T), \quad (4.1)$$

where the growth rate is $F_{\text{N}}(1 - \sigma_{\text{Ga}})$, and N_{Ga} is the number of weakly bound Ga atoms on the substrate surface of area A . Solving Eq. (4.1) for σ_{Ga} at steady state gives the growth rate

$$G = F_{\text{N}}(1 - \sigma_{\text{Ga}}) = F_{\text{N}} \frac{F_{\text{Ga}} - F_0(T)}{F_{\text{N}} - F_0(T)}. \quad (4.2)$$

The result of this model is shown as a solid line in Fig. 9. As stated earlier, we estimate the desorption term $F_0(T_{\text{sub}})$ from the equilibrium vapor pressure of Ga over liquid Ga.²⁶ We see from Fig. 9 that the growth rate as extracted from the Ga data is in reasonable agreement with the steady state solution given in Eq. (4.2). The only fitting parameter used was F_{N} , which is equated to the known Ga flux for which the growth rate is maximum. The main point to be extracted from this discussion is that weakly adsorbed Ga reduces the formation rate of GaN. The model predicts further that the growth ceases completely when $F_{\text{Ga}} = F_0(T_{\text{sub}})$, where $\sigma_{\text{Ga}} = 1$. This

is completely consistent with the results of Sec. 3.5.1, where we found that the onset of Ga condensation is independent of NH_3 flux in the NH_3 -limited regime. Equation (4.2) was also applied to Fig. 8, predicting a deviation from a linear increase in the growth rate as a function of NH_3 flux in the NH_3 -limited regime. This deviation from linearity decreases with increasing substrate temperature due to a reduction in the coverage of weakly adsorbed Ga.

This result shows that we can use substrate temperature to control the coverage of weakly adsorbed Ga. For example, we can increase the growth rate by increasing the substrate temperature, in effect reducing the steady state surface coverage of weakly adsorbed Ga. This is seen in Fig. 10, where the GaN growth rate is plotted as a function of substrate temperature at a fixed Ga and NH_3 flux. The result is counter to the notion that decreased Ga residence times at higher temperature lead to decreased growth rates,¹⁸ as discussed below. We attribute the drop in the growth rate at high temperatures to GaN decomposition competing with growth. This conclusion is motivated by comparing the data to the proposed growth rate model, Eq. (4.2). Two solid curves are shown in Fig. 10; the top one shows the growth rate as given by Eq. (4.2), while the bottom one is given by Eq. (4.2) minus the GaN decomposition rate in the absence of any fluxes. The measured data fall between those two curves, suggesting that GaN decomposition is responsible for the growth rate reduction. Further, during growth, GaN decomposition is suppressed compared to the decomposition measured in the absence of any fluxes.²³

A similar drop in the growth rate was observed by Guha *et al.*¹⁸ at a much lower substrate temperature, and they attributed this reduction to the decrease in the residence time of the incident Ga on the GaN surface. However, their data were apparently taken in the Ga-limited regime, but close to the NH_3 -limited regime, at a Ga flux of 0.6 ML/s and an active N flux of 0.8 ML/s. The substrate temperature ranged from 720°C to 840°C, and the growth rate decreased from 0.5 ML/s at 720°C to zero at approximately 760°C. Those results are in sharp contrast to ours, which were obtained at what should have been very similar growth conditions. Figure 9 indicates that at a Ga flux of 0.6 ML/s and an active N flux of only 0.5 ML/s, at a substrate temperature of 785°C, all of the available Ga incorporates within the

measurement error of less than 0.1 ML/s. Even though our data were taken at a substrate temperature 25° higher, as well as at a lower active N flux, they disagree significantly with Guha's results. Possible differences include errors in substrate temperature determination as well as differences in growth kinetics between NH₃ and RF nitrogen sources.

4.3. Ga-limited growth

The period of the RHEED oscillations shown in Fig. 12 were longer than the growth rates measured by DMS. Further, though the measured periods were linear with the Ga flux, the slope was not unity. This is surprising since it is contrary to experience with most other materials under comparable conditions. Two limiting cases should be considered either the RHEED intensity oscillations correspond to the growth rate but with less than unity Ga incorporation, or the frequency does not correspond to the growth rate, indicating that not all Ga atoms contribute to island nucleation.

Assuming that the RHEED intensity oscillations indicate the true growth rate, we are losing some of the incident Ga by either desorption, scattering or accumulation. In the case of desorption we expect that by increasing the NH₃ flux we can increase the probability of incident Ga to combine with N sites, effectively increasing the slope of the curve. On the other hand, the mean free path of the Ga atoms between the source and the sample shortens at the high NH₃ pressures used, which should result in a decrease in the slope. In either case the constant offset observed is not expected. When increasing the NH₃ flux this offset increased, but the slope of the curve did not change as we might expect. No evidence of Ga accumulation was observed either to explain the behavior.

If we consider the case where the RHEED intensity oscillations do not indicate the true growth rate, we conclude that we are losing some Ga that does not take part in island nucleation. Possibilities might be the migration of Ga atoms to step edges participating in step flow, or the migration to defects on the GaN surface. It becomes evident that a more detailed study is needed to resolve the issue, keeping in mind that several processes combined might be responsible for the observed behavior.

Finally, we should mention that some of the differences obtained between films grown under NH₃-limited conditions and those under Ga-limited conditions might be due to the presence of inversion domains. For example, Romano and Myers²⁷ have recently examined growth on mixed polarity films, using plasma sources for N, and showed that one polarity grows faster than the other under excess N conditions, and that more balanced growth rates are obtained under excess Ga. This resulted in smoother films under excess Ga compared to excess N. We found, however, that even by growing on bulk unipolar GaN(000 $\bar{1}$) surfaces, films showed the 2D and 3D growth regimes. Further, these latter films could be annealed at 800°C, as indicated by a reduction in the intensity of the 3D transmission RHEED features. This suggests that changes in surface diffusion play a dominant role.

4.4. Overall framework

Figure 24 summarizes the findings reported in this paper. The figure shows three main (solid) lines: the incident Ga flux, labeled F_{Ga} , the equilibrium Ga desorption flux, $F_o(T)$, and the flux of Ga corresponding to the decomposition of GaN. In the figure the incident Ga flux is chosen to be $1.1 \times 10^{15} \text{ cm}^{-2} \text{ s}^{-1}$ (1.0 ML/s of GaN). The substrate temperature and the magnitude of the nitrogen flux supplied by the NH₃, F_{N} , determine the growth mode.

For example, if F_{N} and a substrate temperature corresponding to point (a) were chosen, then the growth would be Ga-limited (excess NH₃) and 3D. At somewhat higher NH₃ fluxes, RHEED intensity oscillations would be observed on initially smooth surfaces. Similarly, at point (b), the growth would be Ga-limited and 3D, though at these substrate temperatures the decomposition of GaN becomes significant and reduces the growth rate. In this Ga-limited mode RHEED intensity oscillations are observed, and the observed decrease in oscillation frequency indicates that Ga diffusion can be increased by raising the substrate temperature.

Alternatively, if F_{N} is less than F_{Ga} , then the growth is NH₃-limited (excess Ga) and is described by the data below the horizontal solid line corresponding to the incident Ga flux. The key temperature value corresponds to the intersection of $F = F_o(T)$ and $F = F_{\text{Ga}}$. This is the condensation

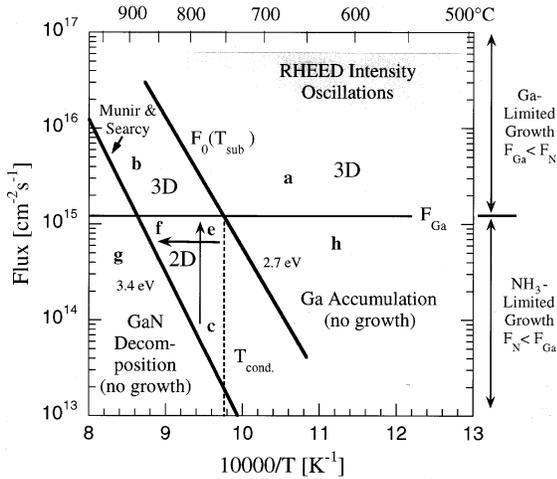


Fig. 24. GaN growth map showing the different growth regimes and limits for a fixed Ga flux. Growth is limited by GaN decomposition at high temperatures and Ga accumulation in the NH_3 -limited regime at low temperatures. 3-D growth occurs in the Ga-limited regime, while layer-by-layer growth takes place in the NH_3 -limited regime between the GaN decomposition and Ga accumulation limits. Strong GaN RHEED intensity oscillations are observed by first growing in the NH_3 -limited regime until a streaky RHEED pattern is obtained, followed by growth in the Ga-limited regime.

temperature, T_c , and is indicated by the vertical dashed line since it is unchanged for all lower NH_3 fluxes at this incident F_{Ga} . For any NH_3 flux in the shaded region to the right of T_c , for example at point (h), Ga will condense on the surface since the incident Ga flux, F_{Ga} , is greater than the flux that would desorb from liquid Ga. There is no GaN growth in this regime. Our results do not show a reduction in desorbed Ga flux in this region when NH_3 is introduced.

In the 2D regime, at temperatures higher than T_c and under excess Ga, growth is possible since multilayers of Ga that would otherwise block growth do not form. In this regime, surface diffusion is evidently high and RHEED oscillations are not observed. We assume that this corresponds to step flow growth. The arrows in this region summarize the results of much of the data presented in Sec. 3 and the model Subsec. 4.2. Under these conditions the $\text{GaN}(000\bar{1})$ surface is terminated by a layer of strongly bound Ga that reacts with NH_3 . However, additional weakly adsorbed Ga blocks growth and reduces the growth rate. The change in growth rate

versus NH_3 flux, shown in Fig. 8, corresponds to the arrow from point (c) to point (e). Along this line the main effect is that the growth rate is limited by the arrival of nitrogen and so increases approximately linearly with NH_3 flux. However, it does not increase linearly with the unity slope since initially there is some fraction of 1 ML of weakly adsorbed Ga on the surface that blocks growth. As the NH_3 flux is increased, the coverage of weakly adsorbed Ga is reduced and hence the fraction of the NH_3 that can be incorporated increases. Similarly, the arrow from point (e) to point (f) is illustrated by the data in Fig. 10. At point (e) there is a fraction of 1 ML of weakly adsorbed Ga on the surface that blocks reaction with the strongly adsorbed Ga. As the substrate temperature is increased, at a fixed Ga and NH_3 flux, the coverage of weakly adsorbed Ga decreases and the growth rate increases. As point (f) is reached, the decomposition of GaN becomes important and there is a reduction in the growth rate. If the substrate temperature were increased past the decomposition line to point (g), there would be no net growth and the decomposition of GaN would dominate.

5. Conclusion

$\text{GaN}(000\bar{1})$ was grown by MBE on sapphire using a Ga Knudsen cell and an NH_3 leak valve, and the growth was investigated by DMS and RHEED. It was shown that DMS and RHEED could be used to observe and control surface termination, Ga coverage and surface temperature. GaN growth and decomposition rates were obtained by DMS. It was shown that hydrogen is present on the surface during growth on $\text{GaN}(000\bar{1})$ and that its desorption rate is proportional to the growth rate. Two distinct growth regimes were identified: growth under excess NH_3 (Ga-limited growth) and growth under excess Ga (NH_3 -limited growth). Under conditions of excess NH_3 , damped oscillations in the specular RHEED intensity were observed. The magnitude of these oscillations depended on sample history and was larger at lower temperatures and higher NH_3 fluxes. Contrary to previous suggestions, the period of these oscillations does not correspond exactly to integral layer deposition and is not characteristic of a narrow growth front. Further, as the substrate temperature was increased the growth mode changed from island nucleation to step flow with an activation

energy of 1.2 eV. When the Ga flux was interrupted, the intensity did not recover without a smoothing step and the growth became 3D. Under conditions of excess Ga, surfaces were much smoother, but RHEED intensity oscillations were not observed, indicating a step flow growth mode. In this regime, it was shown that increasing the Ga flux caused the growth rate to decrease. This reduction was explained by a model that states that weakly adsorbed Ga blocks reactive Ga, which is strongly bound. At low temperatures condensation of Ga limited the growth in the excess Ga regime. It was shown that increasing the substrate temperature reduced the coverage of weakly adsorbed Ga and increased the growth rate. At higher temperatures, growth was limited by GaN decomposition.

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Appendix A: DMS Measurement of the Growth Rate Using Ga

This appendix covers the method used to obtain incorporation data by observing Ga desorption using *in situ* real time DMS.¹¹ So far most of the existing growth rate data on GaN have been obtained by *ex situ* thickness measurements, which, even when a superlattice is grown, require long growth times for each data point. Some *in situ* methods were reported, for example, using the period of RHEED intensity oscillations^{20,19} and IR interference,²⁸ but little growth rate data is published. Under GaN growth conditions of high Ga:NH₃ flux ratios as well as high substrate temperatures, RHEED intensity oscillations were not observed, so that other *in situ* growth rate measurements are required.

We consider deposition on a surface at a temperature above which Ga condenses. A fraction of the incident Ga incorporates as GaN, and the remainder supplies a steady state Ga surface coverage which desorbs at the same rate at which it is supplied. Without any NH₃ flux all the incident Ga desorbs. We assume that an NH₃ flux does not affect the angular distribution of the desorbed Ga flux. In order to obtain the incorporating flux we mea-

sure the difference between these two Ga desorption fluxes.

An example of the Ga desorption signal with and without NH₃ is shown in Fig. 7(a) taken without collimation to improve the signal-to-noise ratio. Here we measure the change in the desorbed Ga signal when the Ga flux is interrupted, since it is somewhat easier to be certain that steady state signals are measured. With NH₃ impinging on the surface, GaN is grown until steady state conditions are obtained. The Ga flux is then interrupted and the DMS signal quickly decreases to a constant background level. We take the drop in the measured mass spectrometer current, ΔI_{growth} , to be proportional to the Ga flux leaving the sample surface. Similarly, without NH₃, there is a drop in the Ga reference signal, ΔI_{ref} , that is proportional to the incident Ga flux. The growth rate is then computed from

$$G = \frac{\Delta I_{\text{ref}} - \Delta I_{\text{growth}}}{\Delta I_{\text{ref}}} \cdot F_{\text{Ga}}, \quad (\text{A.1})$$

using F_{Ga} as measured by GaAs RHEED intensity oscillations and implicitly taking into account the proportionality factor between the measured mass spectrometer current and the desorption flux. Note that at substrate temperatures greater than about 800°C the decomposition of GaN, not included in Eq. (A.1), must also be considered.

Appendix B: Temperature Measurement

The implications of the observations described in the body of the paper are that the Ga condensation temperature on Ga-terminated GaN is a function of the incident Ga flux and that this condensation temperature is determined by the detailed balance of the incident Ga flux and the flux evaporating from liquid Ga. Therefore, to calibrate the substrate thermocouple, we only need to know this incident flux, which can be measured using GaAs RHEED intensity oscillations.¹¹ Using either RHEED (Fig. 20) or DMS (Fig. 6) observations, at a given incident Ga flux we find the lowest temperature at which the incident flux equals the desorbing flux, i.e. the onset of Ga condensation. We then determine the temperature offset of the thermocouple reading by setting the known incident flux equal to the flux corresponding

to the equilibrium vapor, as follows:

$$F_{\text{Ga}} = \frac{p_{\text{eq}}}{\sqrt{2\pi mkT_c}}, \quad (\text{B.1})$$

where p_{eq} is usually tabulated as²⁶

$$\log(p_{\text{eq}}) = A + BT^{-1} + C \log T, \quad (\text{B.2})$$

with p given in units of atm., $A = 6.754$, $B = -13984$ and $C = -0.3413$. Here F_{Ga} is the incident Ga flux and $T = T_c$.

Alternatively, both the incident and desorbing fluxes can be expressed more symmetrically by measuring the incident flux in GaN monolayers per second, $1/\tau_i$, and by fitting the flux desorbing from the surface in the right side of Eq. (B.1) to an exponential over a range of 500–1000°C (also in GaN ML/s). Then Eq. (B.1) becomes

$$\frac{1}{\tau_i} = \frac{1}{\tau_0} \exp\left(\frac{-\Delta H}{k_B T_c}\right), \quad (\text{B.3})$$

where ΔH is the enthalpy of vaporization of Ga and $1/\tau_0$ is an attempt frequency. Hence, at a given incoming Ga flux, the lowest possible surface temperature for avoiding Ga condensation is given by

$$T_c = \frac{\Delta H}{k_B \ln(\tau_i/\tau_0)}, \quad (\text{B.4})$$

where ΔH is 2.71 eV, $\tau_0 = 4.43 \times 10^{-14}$ s/ML, and τ_i is the time it would take to form 1 ML of GaN on a (0001) plane assuming complete Ga incorporation. Thus once the Ga source is calibrated, the surface temperature can be determined by finding the temperature at which Ga condenses, using either RHEED or DMS.

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